

## Analytical Chemistry Experiment (Chemical test material)



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## ANALYTICAL CHEMISTRY EXPERIMENT (CHEMICAL TEST MATERIAL)



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paperback. Book Condition: New. Ship out in 2 business day, And Fast shipping, Free Tracking number will be provided after the shipment. Pages Number: 184 Publisher: China Ocean University Pub. Date :2009-07-01 version 1 2009-07-01 1st printing. WANG Shu-Hao editor of the analytical chemistry experiment in Liaocheng University and other colleges and universities teaching analytical chemistry experiment based on the practice of reform are written. This book focus on theory and practice. in addition to the basic operation of experiments. all experiments begin with the actual samples for the study of more complex objects. while the book also includes qualitative analysis of the contents of the experiment. reflecting the integrity of analytical chemistry. students the ability to apply the knowledge to promote innovation. Contents: Chapter 1 Fundamentals of Analytical Chemistry Experiment 1.1 The purpose of analytical chemistry experiment laboratory water and basic requirements 1.2 specifications. test methods 1.3 Preparation and general knowledge of chemical reagents 1.4 glassware washing. drying and common detergent 1.5 Analytical Chemistry experimental data recording. processing and knowledge of laboratory safety test report 1.6 1.7 waste handling qualitative analysis in Chapter 2 the basics of 2.1 Overview 2.2 qualitative analysis qualitative analysis qualitative analysis of the main instruments 2.3 Chapter 3 of the basic operation of the qualitative analysis of Experiment 1 cation the first working group (Silver Group) - Ag<sup>+</sup>. Pb<sup>2+</sup>. Hg<sup>2+</sup> + analysis experiment 2 cationic group (copper tin set) - Pb<sup>2+</sup>. Bi<sup>2+</sup>. Cu<sup>2+</sup>. Cd<sup>2+</sup>. Hg<sup>2+</sup>. As ( . ). Sb ( . ). Sn ( . ) analysis of Experiment 3 cationic group (iron group) - Al<sup>3+</sup>. Cr<sup>3+</sup>. Fe<sup>3+</sup>. Fe<sup>2+</sup>. Mn<sup>2+</sup>. Zn<sup>2+</sup>. Co<sup>2+</sup>. Ni<sup>2+</sup> + analysis experiment 4 cations calcium and sodium fine - intoxicated + aim +. commanded ten. I (+). Nla...



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