

## The Position of Lime in the Chemistry of the Soil (Classic Reprint) (Hardback)



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**(Prof. Juliana Langosh DVM)**

## THE POSITION OF LIME IN THE CHEMISTRY OF THE SOIL (CLASSIC REPRINT) (HARDBACK)



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Forgotten Books, 2017. Hardback. Condition: New. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*. Excerpt from The Position of Lime in the Chemistry of the Soil But if we are to correctly advise our agricultural trade, then we must understand our subject thoroughly. It is broad and complex and must be studied diligently. Still it is founded on a few fundamental facts, which are exceedingly simple. It is these facts and the position of lime in reference to them with which we are here dealing. It must first be understood that there are eight necessary ingredients to plant life. One authority, whom we will accept states. Soils contain nitrogen, phosphorus, potassium, calcium, magnesium, sulphur, iron and chlorine, and if a soil is deficient in any of these essential elements of plant food, it will not produce a satisfactory crop. The first four are of major importance, the last four are minor. Of course, it is well known that other things are needed such as soil-air, moisture and humus or decaying organic matter, but we are dealing merely with the mineral chemistry of the soil and plant life. Also plants need carbon for their structure and this is obtained both from the atmosphere and from the humus of the soil, which in decaying evolves the gas, carbon dioxide or carbonic acid gas. These elements do not appear in the soil or plant in their elementary form, but always as compounds. It is to be noted that of these substances named above, some are acid-forming and others are bases, i. e. form alkalis. The acid-forming are nitrogen, phosphorus, sulphur and chlorine, becoming respectively, nitric, phosphoric, sulphuric and hydrochloric acids. Carbon is also acid, forming carbonic acid. The bases are potassium, calcium and magnesium forming potassium, hydroxide, etc. The chemical...



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